Additive- and Ligand-Free Cross-Coupling Reactions between Alkenes and Alkynes by Iridium Catalysis

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ABSTRACT: Although a range of transition-metal-catalyzed cross-coupling reactions of alkenes and alkynes have been developed to produce valuable conjugated dienes, extension of these reactions to iridium catalysis has yet to be demonstrated. The first iridium-catalyzed alkene−alkyne cross-coupling reactions have been realized under ligand- and additive-free conditions. A wide range of acrylamides and alkynes could be used as coupling partners, providing branched (Z,Z) butadiene skeletons with excellent site- and stereoselectivities.

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ross-coupling reactions are highly sought after in pharmaceutical and materials chemistry, converting readily available raw feedstocks to useful, value-added commodity chemicals.^{[1](#page-3-0)} In particular, the formation of C(alkenyl)−C(alkenyl) bonds via oxidative C−H/C−H cross-coupling is highly desirable because conjugated diene skeletons are ubiquitous in natural products and drugs. $²$ $²$ $²$ </sup>

The catalytic alkene−alkyne coupling reactions provided an attractive access from the standpoint of atomic economy, which has been achieved via metallacyclopentene, β chelation-assisted hydrovinylation of alkyne,^{[2i,4](#page-3-0)} or a Mizoroki-Hecktype mechanism, 5 usually leading to a Z,E-configuration. A central challenge in this area is the development of divergent methods to achieve highly regio- and stereoselective formation of either branched or linear butadienes. New catalytic protocols with additive-free, ligand-free, and/or oxidant-free conditions, broader substrate scope, better reactivity, as well as unique chemo- and stereoselectivity are still highly desirable. Transition metals of the fourth and fifth period, such as ruthenium, rhodium, palladium, nickel, and cobalt, have been utilized in alkene−alkyne cross-coupling reactions.^{[3](#page-3-0)−[5](#page-3-0)} However, there is still no example catalyzed by transition metal of the sixth period such as iridium.^{[6](#page-3-0)}

In light of the rapid development of Ir-catalyzed C−H functionalizations, 7^{-9} 7^{-9} 7^{-9} 7^{-9} it is highly desirable to extend the alkyne−alkene cross-couplings to iridium chemistry for a number of reasons. First, both the substrate scope and reaction type can be complementary to previous reactions or even with better reactivity and selectivity.^{[7](#page-3-0)} Second, the different ligands used for Ir-catalyzed C−H functionalization could offer new opportunities for developing ligand-controlled, chemo- and stereoselective cross-coupling reactions.^{[9](#page-4-0)} Despite remarkable progress on Ir-catalyzed aromatic and alkyl C−H functionalizations,[7](#page-3-0),[8](#page-4-0) examples of alkenyl C−H functionalization by iridium catalysis continue to be scarce, 10 presumably due to lability of alkenes under highly active iridium catalysis. Moreover, competitive C(alkenyl)−H and C(allyl)−H cleavage sites, displaying ready olefin isomerization via π allyliridium or iridium hydride intermediate, $7,10,11$ $7,10,11$ $7,10,11$ $7,10,11$ can make reactivity and selectivity of cross-coupling reaction a significant challenge.

Using ubiquitous acrylic acids or their derivatives such as acrylamides as coupling partners is advantageous due to their low cost and widely occurring in natural products and drugs (Scheme 1b). In line with our ongoing interest in olefinic C− H activation and butadiene synthesis,^{[2g](#page-3-0)-[i,4e](#page-3-0)} we herein disclose Ir-catalyzed regio- and stereoselective cross-couplings using structurally diverse acrylamides and alkynes, leading to branched Z,Z-configured butadienes (Scheme 1a). The utility

Received: May 20, 2019 Published: June 12, 2019

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of this approach is also demonstrated by preparative-scale and C−H functionalization of perillic and artemisic amides.

Due to the widely occurring and great importance of amide group, we examined the cross-coupling between different Nsubstituted acrylamides 1 and alkyne 2a. Unfortunately, Nalkylamides led to poor reactivity/selectivity under a variety of catalytic conditions. In this context, we envisioned that high acidity of the N−H proton might be essential for the formation of the amidoiridium species via deprotonation to facilitate the C−H insertion.[7](#page-3-0) We turned to examine the reaction of acrylamide bearing a p-toluenesulfonyl group. Although $[IrCp*Cl₂]$, was inefficient, the cross-coupling reaction was found to take place in the presence of $[\text{Ir}(\text{OMe})(\text{cod})]_2$ (2) mol %), yielding product 3aa in 74% yield (Table 1, entries 1

Table 1. Optimization of Catalytic Conditions^a

NHTs O. Me 1a	$\ddot{}$ $Ph-$ -Ph 2a	[lr]	TsHN. Me	Ph Ph Заа
entry	catalyst	additive	solvent	yield ^b $(\%)$
1	$[\text{IrCp*}Cl_2]_2$		toluene	θ
\mathfrak{p}	$[\text{Ir}(\text{OMe})(\text{cod})],$		toluene	74
3	[IrCl(cod)],		toluene	87
$\overline{4}$	[IrCl(cod)],		dioxane	86
5	[IrCl(cod)],		DCE	83
6	[IrCl(cod)],		DME	71
7	[IrCl(cod)],		MeOH	96
8	[IrCl(cod)],		EtOH	72
9	[IrCl(cod)],		$P_{r}OH$	76
10	[IrCl(cod)],		H ₂ O	73
11 ^c	$[\text{IrCl}(\text{cod})]_2$	AgSbF ₆	MeOH	85
12^c	[IrCl(cod)],	AgOTf	MeOH	87
13^d	[IrCl(cod)],		MeOH	84

^aReaction conditions: 1a (0.2 mmol), 2a (0.3 mmol), [Ir] (4 mol %), in solvent (1 mL) at 70 °C, 24 h. b^b Isolated yields. ^cAg additive (4 mol
%) added. ^d1 mol % of $[\text{IrCl(cod)]}_2$ used, 48 h. DME = 1,2dimethoxyethane; DCE = 1,2-dichloroethane.

and 2). This prompted us to try another catalyst $[\text{IrCl}(\text{cod})]_2$ also chelated by diene ligand, leading to 87% yield (entry 3). Replacing the toluene with dioxane, DCE, or DME did not improve the reaction (entries 4−6). To our delight, using of MeOH led to quantitative yield and excellent stereoselectivity $(Z,Z/Z,E > 99/1)$ (entry 7). The Z,Z-configuration supported a directed oxidative addition of the olefinic C−H bond followed by a syn-insertion into the alkyne. 4 Other alcohols such as ethanol and 2-propanol led to decreased yields (entries 8 and 9). Interestingly, the reaction proceeded even in water, albeit with decreased efficacy (entry 10). Notably, addition of silver salts such as $AgSbF_6$ and $AgOTf$ even retarded the reaction slightly (entries 11 and 12).¹² The alkenylation still proceeded well under lower catalyst loading (entry 13).

With the optimized reaction conditions in hand, we next examined the scope and limitation of alkenes 1 and alkynes 2 (Scheme 2). The reactions of dialkylacetylenes 2b and 2c also reacted well with acrylamide 1a to give the Z,Z-configured products 3ab and 3ac in 69% and 77% yields, respectively. Diarylacetylenes bearing Br or alkyl substituents proceeded smoothly to give the products in excellent yields (3ad and 3ae). In the reactions of unsymmetrically substituted alkynes such as 1-phenyl-1-propyne, C−C bond formation only

^aReaction conditions: 1 (0.2 mmol), 2 (0.3 mmol), $[\text{IrCl(cod)}]_2$ (2 mol %) in MeOH (1 mL) at 70 °C for 24 h. $b^{\text{Isolated yields.}}$ °Z,Z/ $Z,E = 87/13$.

occurred at the C-2 position, indicating excellent regioselectivity governed by electronic and steric factors (3af). Unfortunately, both bis(2-methylphenyl) acetylene and phenylpropiolate failed to react even at elevated temper-atures.^{[13a](#page-4-0)} Thereafter, we turned to the reactions of other representative aliphatic and aryl substituted acrylamides 1. Aromatic F, Cl, CF_{3} , and OMe were also well tolerated and gave the corresponding products in 67−98% yields, thus showing good functionality compatibility (3ba−3ga). 6- Dodecyne also reacted well, leading to 93% yield (3bg). Notably, acrylamide bearing a sensitive benzyl group was also converted (3ha). Installation of longer alkyl chain at the α position furnished the corresponding cross-coupling in excellent yields (3ia and 3ja). Even angelic amide and plain acrylamide were converted, though both of them were totally inert or incompatible in previous directed C(alkenyl)−H functionalizations (3ka and 3la).^{[2](#page-3-0),[4](#page-3-0)} Conformationally restricted acrylamide bearing a cyclohexenyl or a cyclopentene moiety converted smoothly (3ma and 3na). All of these results exhibited a much broader substrate scope than the previous Ru-catalyzed protocol.^{[4e](#page-3-0)} Finally, N-methylsulfonyl acrylamide 1o gave the cross-coupling product 3oa in 87% yield.^{[13b](#page-4-0)}

Given the high catalytic efficacy of the iridium-catalyzed cross-couplings, we attempted to gain some preliminary understanding of the reaction mechanisms. Competition experiments between alkynes 2a and 2g highlighted alkylalkynes to be more reactive ([Scheme 3a](#page-2-0)). Intermolecular competition experiments between acrylamides 1d and 1c revealed that electron-deficient acrylamide was converted preferentially, hence rendering an electrophilic C−H bond activation less likely to be operative ([Scheme 3](#page-2-0)b).^{[2](#page-3-0)} These

Scheme 3. Competitive Reactions

results are consistent with our prior observation in rutheniumcatalyzed alkene−alkyne coupling.[4e](#page-3-0)

If acrylamide 1b was subjected to the corresponding optimal conditions with CD_3OD , significant olefinic H/D exchanges were observed, thereby implicating a reversible cyclometalation mode (Scheme 4a). The excellent site- and stereoselective H/

D exchanges excluded the possible alkyliridium mechanism by insertion.⁷ Treatment of deuterium-labeling acrylamide 1b- \vec{d} , with acetylene 2a in protic methanol led to a complete deuterium loss in the olefinic position of product 3ba-d, while the cis-olefinic deuterium completely transferred to the product $3ba-d_2$ using DCE as a solvent (Scheme 4b). Although a number of deuteration of alkenes have been reported, extension to butadiene derivatives continued to be scarce, and such selective deuteration may find valuable utilization in mechanistic and metabolic studies. 14 We also performed the deuterium incorporation experiment under the optimal conditions in the presence of alkyne 2a. The lack of Dincorporation to recovered acrylamide 1b suggested that the alkenylation step is sufficiently fast to outcompete the reversibility of the C−H activation step, while 21% Dincorporation to the product 3ba exhibited a fast H/D exchange on hydridoiridium intermediate (Scheme 4c). Finally, kinetic isotope effect (KIE) values determined via intermolecular competition experiments and parallel experiments using non- and deuterated acrylamides 1b and $1b-d_2$ confirmed olefinic C−H cleavage to be the rate-determining step (Scheme 4d).¹⁵

The gram-scale cross-coupling reaction also proceeded well even with 1 mol % catalyst loading, as illustrated by the preparation of 3aa to demonstrate the robustness of the protocol (Scheme 5a). The amide group can be conveniently removed under methylation−hydrolysis conditions, providing

Scheme 5. Synthetic Applications

carboxylic acid 5 without any erosion of the Z/E configuration (Scheme 5b). A particularly useful application of this method is in the late-stage C−H derivatization of bioactive molecules such as perillic and artemisic acid derived amides. All of them reacted smoothly, leading to alkenylation analogues 3pa−3qa in 71−96% yields with excellent site- and stereoselectivities (Scheme $5c$).^{[16](#page-4-0)} Given that many therapeutic agents contain acrylic acid/acrylamide moieties, we expect our methodology to provide valuable opportunities to accelerate structure− activity relationship studies in drug discovery.

Plausible catalytic cycles are shown in Scheme 6. First, [Ir]- Cl reacts with N-sulfonyl acrylamide 1 to give an amidoiridium

Scheme 6. Proposed Mechanism

species A1. Oxidative addition of olefinic C−H bond to the iridium produces a hydridoiridium intermediate A2, which reacts quickly with alkyne 2 by syn-addition to generate the intermediate A3. Reductive elimination and ligand exchange by amide 1 give the diene 3 and regenerate A1 (Scheme 6).^{7,[14](#page-4-0)}

In summary, we have presented the first cross-couplings of alkenes and alkynes by iridium catalysis, leading to stereoselective synthesis of a wide range of branched Z,Z-dienamide derivatives. By judicious choice of iridium complexes [IrCl- (cod) ₂, the cross-couplings proceeded under metal additiveand ligand-free conditions, leading to good to excellent reactivity and selectivity. The operationally simple protocols displayed broad substrate classes, including mono-, di-, and trisubstituted acrylamides and enabled the gram-scale prepara-

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tion and C−H modification of perillic and artemisic amides, demonstrating their practicality and versatility. Furthermore, the N-Ts amide auxiliary could be smoothly removed under mild conditions. We anticipate that these divergent crosscoupling reactions will find broad applicability in multifarious synthetic endeavors.

■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the [ACS Publications website](http://pubs.acs.org) at DOI: [10.1021/acs.or](http://pubs.acs.org/doi/abs/10.1021/acs.orglett.9b01766)[glett.9b01766.](http://pubs.acs.org/doi/abs/10.1021/acs.orglett.9b01766)

Experimental procedures and spectral data for all new compounds [\(PDF](http://pubs.acs.org/doi/suppl/10.1021/acs.orglett.9b01766/suppl_file/ol9b01766_si_001.pdf))

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Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We gratefully acknowledge the National Natural Science Foundation of China (NSFC) (21502037 and 21672048), the Natural Science Foundation of Zhejiang Province (ZJNSF) (LY19B020006), and the Hangzhou Normal University for financial support. G.Z. acknowledges a Qianjiang Scholar award from Zhejiang Province, China.

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(13) (a) Terminal alkynes showed no reactivity toward acrylamide. (b) NH-Me acrylamide exhibited limited reactivity (∼15% yield), while N-Me-N-Ts acrylamide exhibited no reactivity.

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previous coupling conditions (ref [4e](#page-3-0)) presumably due to the presence of silver salt.